SYNTHESIS OF [4.5.5.5] FENESTRANE AND A [4.4.5.5] FENESTRANE DERIVATIVE

William G. Dauben* and Daniel M. Walker Department of Chemistry, University of California, Berkeley, CA 94720

<u>Abstract</u>: The synthesis of tetracyclo $[5.4.1.0^{4,12}.0^{9,12}]$ dodecan-6-one 12 via an intramolecular photocycloaddition, its reduction to the hydrocarbon 13 and its ring-contraction to tetracyclo- $[4.4.1.0^{3,11}.0^{9,11}]$ undecane derivative 16 is described.

The hydrocarbon tetracyclo[$3.3.1.0^{3,9}.0^{7,9}$]nonane 1, known as [4.4.4.4]fenestrane, or simply windowpane, has been the subject of several recent discussions¹ and synthetic studies.² The synthetic work has resulted in the synthesis of [5.5.5.5]fenestrane derivatives 2^{2b} and 3^{2c} and the [4.4.4]fenestranes 4^{2e} and 5^{2f} . In addition to the overall aesthetic appeal of windowpane, the central quaternary carbon is of particular interest. This carbon, although



not constrained to planarity, is expected to be severely distorted from the normal tetrahedral geometry. It has been estimated that the $C_1-C_9-C_5$ angle in 1 should be 140° and the strain at the central carbon would be roughly 40 kcal/mol.^{1b} This degree of distortion and strain is unprecedented and thus the stability and reactivity of 1 is difficult to predict. We have therefore embarked on a stepwise program designed to synthesize the homologous series of [4.5.5.5], [4.4.5.5], [4.4.4.5] and [4.4.4.4]fenestranes and to study the effects of increasing distortion at the central carbon on the properties of these compounds.

Our basic synthetic approach (Scheme I) is to use an intramolecular photocycloaddition of a bicyclic enone with an olefin to generate simultaneously the central quaternary carbon and the final two rings of the fenestrane.³ Subsequent ring contractions of the initial photoadduct could then be used to generate lower and more strained homologues. We would herein 712



like to describe the successful application of this strategy to the first synthesis of [4.5 5.5] fenestrane and a [4.4.5.5] fenestrane derivative.

The synthesis of the requisite 8-substituted bicyclo[3.3.0]oct-1-en-3-one was achieved as follows (see Scheme II). The potassium salt of 2-carbomethoxycyclopentanone was alkylated with 4-bromo-1-butene (1.1 eq, β CH₃, cat. 18-crown-6, KI, reflux, 24 h) to yield β -keto ester p^4 in 95% yield. Reaction of p with methanolic sodium methoxide (1 eq, reflux, 13 h) followed by azeotropic removal of methanol with toluene effected a retro-Dieckmann reaction and subsequent recyclization to the stable β -keto ester enolate. This enolate was alkylated with ethyl bromoacetate(1.3 eq, ØCH₃, reflux, 21 h) to afford a 79% yield of dialkylated cyclopentanone 7.4,5 The alkylation-directing carbomethoxy group was efficiently removed by S_N^2 ester cleavage and decarboxylation⁶ (4 eq anhyd. LiCl, 1.1 eq HOAc, HMPT, 110°C, 17.5 h) to give a cis-trans mixture of 2,5-disubstituted cyclopentanone g^4 (82%). Protection of the carbonyl was effected in quantitative yield to give 2^4 (cis/trans = 33:67) by treatment of 8 with ethylene glycol (5 eq), triethylorthoformate (2 eq) and p-toluenesulfonic acid (0 06 eq, 25°C, 25 h). The ester 9 was added to lithium dimethyl methylphosphonate (2.5 eq, THF, -78°C → 25°C, 5 h) and the reaction was quenched with 2N HCl (25°C, 2.5 h) to afford the β , ϵ -diketophosphonate 10^4 in 89% yield from 7.⁷ The critical intramolecular Wadsworth-Emmons cyclization was effected by a modification of the procedure of Aristoff⁸ which afforded an excellent yield of enone 11 and epimerized the butenyl side chain to the more stable exo configuration without migration of the cyclopentanone double bond. Thus, treatment of 10 with anhydrous K_2CO_3 (2 eq) and 18-crown-6 (0.3 eq) in THF (40°C, 21 h) afforded enone $11^{4,9}$ in 92% yield as a 95:5 mixture of exo-endo isomers. The intramolecular photocycloaddition was smoothly accomplished by irradiation of 11 with a 450 W Hanovia (hexanes, 10^{-2} M, 25°C, 68 h) through a uranium glass filter ($\lambda > 330$ nm) to give the [4.5.5.5] fenestrone $12^{4,10}$ in 95% yield Wolff-Kishner reduction of ketone 12 (NH₂NH₂·H₂O, N(CH₂CH₂OH)₃, KOH, 210°C, 5 h) afforded the volatile [4.5.5.5] fenestrane $13, 4, 1\overline{1}$ which was isolated in 71% yield via preparative GLC The ¹³C spectrum of 12 displayed seven lines, in full accord with the C₂ symmetry possessed by this molecule.

It has been shown that the photochemical Wolff-rearrangement of α -diazo ketones derived from bicyclo[3 2.0]heptan-2-ones is an efficient ring contraction route to bicyclo[2.2.0]hexane-2-carboxylates.¹² Therefore, ketone 12 was treated with sodium hydride (1.2 eq) and ethyl formate (2.6 eq, Et₂0, 0°C \rightarrow 25°C) to afford the crude α -formyl ketone 14 Reaction of 14 with tosyl azide (1 eq) and triethylamine (2 eq, CH₂Cl₂, 25°C, 4 h) gave α -diazo ketone 15.¹³ The crude α -diazo ketone 15 was irradiated with a 450 W Hanovia through Pyrex (MeOH, 10⁻² M, 0°C, 2 h) to afford a 61·39 mixture of epimeric methyl esters 16^{4,14} in 61% yield from ketone



 $\frac{12}{12}$. Epimerization of the mixture in refluxing methanolic sodium methoxide gave a thermodynamic ratio of 53.47 with the same epimer predominating. The fenestrane 16 does not appear to have any unusual instability and has been fully characterized as the inseparable epimeric mixture.

Thus, the synthesis of a derivative of tetracyclo $[4.4.1.0^{3,11}.0^{9,11}]$ undecane, $\frac{16}{\sqrt{5}}$, the smallest and most strained [m.n.p.q] fenestrane known to date, has been achieved in 11 steps in a 29% overall yield from dimethyl adipate.

Acknowledgment. This work was supported by National Science Foundation Grant No. CHE-7804811.

References and Notes

- a) A. Greenberg and J. F. Liebman, "Strained Organic Molecules," Academic Press, N. Y, 1978, Chap. 6. b) K. B. Wiberg, G. B. Ellison and J. J. Wendolski, J. Am. Chem. Soc. 1976, 98, 1212. c) V. I. Minkin, R. M. Minyaev and V. I. Natanzon, <u>Zh. Org Khim</u>, 1980, 16, 673. d) D. C. Crans and J. P. Snyder, J. Am. Chem. Soc. 1980, 102, 7152. e) E.-U Wurthwein, J. Chandrasekhar, E. D. Jemmis and P.v.R. Schleyer, <u>Tetrahedron Lett</u>. 1981, <u>22</u>, 843.
- a) V. Georgian and M. Saltzman, <u>Tetrahedron Lett</u>. 1972, 4315. b) R. Mitschka, J. M. Cook and U. Weiss, <u>J. Am. Chem. Soc</u>. 1978, 100, 3973. c) R. Keese, A. Pfenninger and A. Roesle, <u>Helv. Chim. Acta 1979, 62</u>, 326. d) W. Ten Hoeve and H. Wynberg, <u>J. Org. Chem</u>. 1980, 45, 2925, 2930. e) K. B. Wiberg, L. K. Olli, N. Golembeski and R. D. Adams, <u>J. Am. Chem. Soc</u>. 1980, 102, 7467. f) S. Wolff and W. C. Agosta, <u>J. Chem. Soc., Chem. Commun</u>. 1981, 118.
- 3. This approach has been successfully employed in the synthesis of [4.5.5.6] and [4.5.6.6]fenestrane derivatives. See ref. 2a.
- This compound has been fully characterized by spectral means and gave a satisfactory C and H combustion analysis.
- 5. K. Sisido, K. Utimoto and T. Isida, J. Org. Chem. 1964, 29, 2781.
- 6. Under strictly anhydrous conditions, the acetic acid is essential to protonate the intermediate enolate anion thus preventing alkylation by the methyl chloride generated in the $S_N 2$ reaction. P. Müller and B. Siegfried, <u>Tetrahedron Lett</u>. 1973, 3565. For a general review of the $S_N 2$ ester cleavage reaction see J. McMurry, <u>Org. React</u>. 1976, <u>24</u>, 187
- W. G. Dauben, G. Ahlgren, T. J. Leitereg, W. L. Schwarzel and M. Yoshioka, <u>J. Am Chem Soc</u>. 1972, <u>94</u>, 8593; R. D Clark, L. G. Kozar and C. H. Heathcock, <u>Synth. Commun</u> <u>1975</u>, <u>5</u>, 1, M. J. Begley, K. Cooper and G. Pattenden, <u>Tetrahedron Lett</u>. <u>1981</u>, <u>22</u>, 257.
- 8 P A. Aristoff, J. Org. Chem. 1981, 46, 1954. For a general review of the intramolecular Wittig reaction see: K. B. Becker, <u>Tetrahedron 1980</u>, 36, 1717.
- 9. UV (cyclohexane) λ_{max} 225 nm (ϵ 13100), 302 (31), 311 (35), 321 (35), 334 (26), 352 (10).
- 10. ¹H-NMR (250 MHz, CDCl₃) δ 1.10-1.38 (m,2), 1.38-1.55 (m,1), 1 55-1.73 (m,1), 1.75-1 95 (m,2), 2.03-2.45 (m,8), 2.51 (dd, 1, J = 4.7, 10.0), 2.73 (dd, 1, J = 7.7, 17 2)
- 11. 13_{C-NMR} (CDC1₃) δ 27.8 (t, J = 133), 33.8 (t, 2C, J = 128), 34.5 (t, 2C, J = 127), 35.3 (t, 2C, J = 127), 42.2 (d, 2C, J = 135), 51.2 (d, 2C, J = 130), 70.6 (s).
- K. B. Wiberg, B. L. Furtek and L. K. Olli, <u>J. Am. Chem. Soc</u>. <u>1979</u>, <u>101</u>, 7675. See also ref. 2e and 2f.
- 13. M. Regitz and J. Rüter, Chem. Ber. 1968, 101, 1263.
- 14. 1 H-NMR (250 MHz, CDCl₃) δ 0.88-1.12 (m,2), 1.53-1.74 (m,2), 1.83-2.45 (m,8), 2 66-2.80 (m,2), 3.06 (dd, 0.6H, J = 4.3, 8.5), 3.23 (dd, 0.4H, J = 3.6, 7.9), 3.65, 3.70 (2s, 3 (40:60)), 1³C-NMR (CDCl₃, minor isomer in parentheses) δ 27.3, (31.6), (34.4), (34.9), 35.36, (35.40), 35.6, (36.37), 36.44, (39.45), 39.53, 40.4, 40.6, (40.7), 42.0, (43.7), (45.2), 45.6, 49.0, (49.2), (51.08), 51.13, (63.5), 64.6, (174.2), 174.4.

(Received in USA 2 November 1981)